

## Chapter 9 Mastering Chemistry Answers

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ANSWER: ANSWER: 3.6 mol of Fe<sub>2</sub>O<sub>3</sub> 0.9 mol of Fe<sub>2</sub>O<sub>3</sub> 2 mol of Fe<sub>2</sub>O<sub>3</sub> 1.8 mol of Fe<sub>2</sub>O<sub>3</sub> The amount of is the amount of. four times one-quarter twice one-half equal to Typesetting math: 75% 3/31/2019 Ch 9 HW 3/15 Correct Part B How many moles of hydrogen gas would be needed to react with excess carbon dioxide to produce 88.1 of water vapor?

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Problem 9.113 Assume that you have 1.95 g of nitroglycerin in a 605.5 mL steel container at 20.0 °C and 1.00 atm pressure. An explosion occurs, raising the temperature of the container and its contents to 425 C.

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appropriate atom(s) by selecting each atom and assigning it a map number of 1 until atoms are mapped. To do this, right click on an atom and choose Atom Properties.

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Mastering Chemistry Chapter 13 Carbohydrates Course Home 9 of 21 Review Constants Periodic Table CH,  $\overset{\text{O}}{\text{C}}$  CH-CH-CH-CH<sub>3</sub> Identify the appropriate atom(s) by selecting each atom and assigning it a map number of 1 until atoms are mapped. To do this, right click on an atom and choose Atom Properties. (Mac users: Use an equivalent for right-clicking.)

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