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phosphorus in the second of these three
reactions The answer is 250×10^4 kg P
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SECTION 3 SHORT ANSWER Answer the
following questions in the space
provided. 1. Match description on the
right to the correct crystal type on the
left. b ionic crystal (a) has mobile

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Answers

electrons in the crystal c covalent
molecular crystal (b) is hard, brittle, and
nonconducting a metallic crystal (c)
typically has the lowest melting point of
the four

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Chapter 10 Chemical Calculations and
equations 10, that can be made from
 109×10^4 kilograms of phosphorus in
the second of these three reactions The
answer is 250×10^4 kg P₄O₁₀ We
used the following steps:

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Name SECTION 2 Date CHAPTER 11

REVIEW Gases Class SHORT ANSWER

Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in

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because an ionic compound: Holt
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Quiz DRAFT. 10th - 12th grade. 0 times.
Chemistry. 0% average accuracy. 18
minutes ago ...

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The number 6.022×10^{23} , called

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Avogadro's number after the 19th-century chemist Amedeo Avogadro, is the number we use in chemistry to represent macroscopic amounts of atoms and molecules. Thus, if we have 6.022×10^{23} Oxygen atoms, we say we have 1 mol of Oxygen atoms.

CH105: Chapter 1 - Measurements

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Answers in Chemistry - Chemistry

Modern Chemistry 6 The Periodic Law
CHAPTER 5 REVIEW The Periodic Law
SECTION 3 SHORT ANSWER Answer the
following questions in the space
provided. 1. _____ When an electron is
added to a neutral atom, energy is (a)
always absorbed.

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